# IRON(III) CLUSTERS WITH 3-FORMYLSALICYLIC ACID 

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In the presence of additional ligands and polar solvents, the well-known $\mu_{3}$-oxo bridged triangular iron carboxylates easily reorganize and stabilize large complexes. We chose 3 -formylsalicylic acid $\left(\mathrm{H}_{2} \mathrm{~L}\right)$ in order to explore its coordinative ability, due to its multiple coordination donor groups like $-\mathrm{COOH},-\mathrm{OH}$ and -CHO Herein, new hexanuclear iron(III) clusters $\left[\mathrm{Fe}_{6} \mathrm{O}_{2}(\mathrm{OH})_{2}(\mathrm{~L})_{2}\left(\mathrm{H}_{2} \mathrm{O}\right)_{2}\left(\mathrm{O}_{2} \mathrm{CR}\right)_{8}\right] \cdot \mathrm{MeCN}\left(\mathrm{R}=\mathrm{CMe}_{3}(\mathbf{1})\right.$; $\mathrm{CHMe}_{2}$ (2)), where L is the dianion of 3-formylsalicylic acid, are reported. Compounds $\mathbf{1}$ and 2 were prepared from the reaction of $\mu_{3}$-oxo trinuclear iron(III) precursors $\left[\mathrm{Fe}_{3}\left(\mu_{3}-\right.\right.$ $\left.\mathrm{O})\left(\mathrm{H}_{2} \mathrm{O}\right)_{3}\left(\mathrm{O}_{2} \mathrm{CR}\right)_{6}\right]\left(\mathrm{O}_{2} \mathrm{CR}\right) \cdot 2 \mathrm{HO}_{2} \mathrm{CR}$ with 3-formylsalicylic acid in acetonitrile. Both structures comprise six Fe atoms in an almost planar arrangement that can be described as two oxo-centered triangular units $\left[\mathrm{Fe}_{3}\left(\mu_{3}-\mathrm{O}\right)\right]^{7+}$ joined together by two bridging hydroxide and two bridging carboxylate groups. The asymmetric units contain only half of an $\mathrm{Fe}_{6}$ molecule. All Fe atoms adopt distorted octahedral coordination geometries and are in the +3 oxidation state. The peripheral ligation of metal ions is completed by six carboxylate molecules and two aldehyde ligands which are in their monoanionic and dianionic forms, respectively. The two 3 -formylsalicylic ligands act as tridentate, bridging Fe 1 ( $\mathrm{Fe} 1^{\prime}$ ) and Fe 2 ( $\mathrm{Fe} 2^{\prime}$ ) ions by the alkoxo groups within each $\left[\mathrm{Fe}_{3}\left(\mu_{3}-\mathrm{O}\right)\right]^{7^{+}}$ unit. The oxigen atom of the formyl group is also coordinated, which was not reported so far for this ligand. All carboxylate groups adopt the bridging $\mu_{2}-\eta^{1}: \eta^{1}$ coordination mode: two of them join the edges of the triangular $\left[\mathrm{Fe}_{3}\left(\mu_{3}-\mathrm{O}\right)\right]^{7+}$ units and the remaining six link Fe atoms within the latters (Fig. 1). Detailed characterization of compounds $\mathbf{1}$ and $\mathbf{2}$ will be described in a forthcoming paper.



Figure 1. The molecular structure of complex
2. $\mathrm{CH}_{3} \mathrm{CN}$ (left) and its $\left[\mathrm{Fe}_{6}\left(\mu_{3}-\mathrm{O}\right)_{2}(\mu-\mathrm{OH})_{2}\right]^{12+}$ structural core (right).

Color code: $\mathrm{Fe}^{\mathrm{III}}$, olive green; O , red; C , grey. The hydrogen atoms, methyl groups and solvent molecules are omitted for clarity.

Keywords: 3-formylsalicylic acid, asymmetric units, hexanuclear iron(III) clusters, peripheral ligation.

